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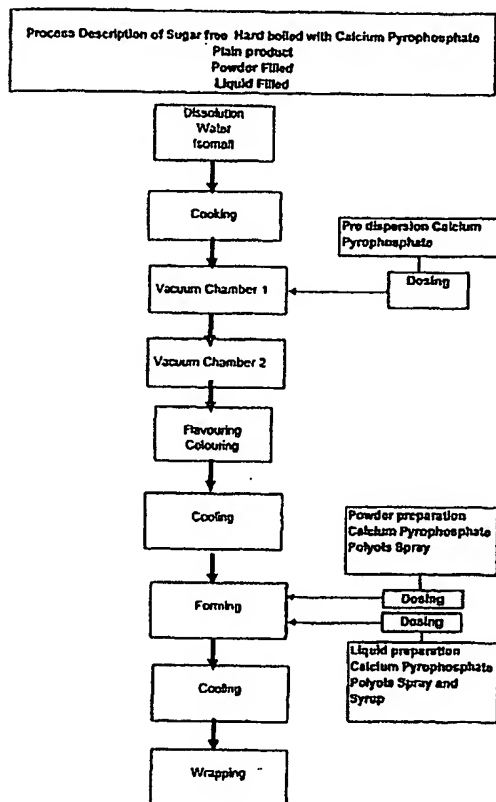
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[Continued on next page]

(54) Title: A SOLID ORAL TOOTH WHITENING CONFECTIONARY COMPOSITION



(57) Abstract: The present invention provides a solid, oral tooth whitening confectionary composition comprising a tooth whitening agent comprising an alkaline or alkaline earth metal pyrophosphate, preferably calcium pyrophosphate. In one embodiment, the composition of the invention comprises an additional tooth whitening agent.

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Title: A solid oral tooth whitening confectionary composition

Technical Field

- 5 The invention relates to solid, oral tooth whitening confectionary compositions. The invention further relates to the use of such compositions to whiten tooth surfaces.

Background Art

- 10 Tooth whitening or stain removing agents are known to be added to dentifrice compositions such as toothpaste, mouthwash, chewing gum, confectionary compositions and the like. The use of such compositions for reducing stains and discolouration of tooth surfaces thereby improving the general cosmetic appearance of the teeth is likewise well-known. Teeth with extrinsic stains are objectionable both on the basis
15 of cosmetic appearance and also socially as indication of poor oral hygiene.

- Some products contain peroxides, but these are, however, problematic from a toxicological point of view. Another approach to tooth whitening products is to add abrasives - known mainly from dentifrices. Not all of these are legal in confection-
20 ary. Further, a significant tooth whitening effect would not be expected to occur following the consumption of confectionary compositions comprising abrasives as these compositions are not suitable for continuous chewing.

- Several abrasive agents have been used for tooth whitening purposes and these are
25 known to the person skilled in the art. Examples of abrasive agents include calcium carbonate, sodium bicarbonate, sodium metaphosphate, potassium metaphosphate, tricalcium phosphate, dihydrated dicalcium phosphate, calcium pyrophosphate, bentonite, zirconium silicate or other siliceous materials. Other suitable abrasives are described in U.S. Patent No. 4,170,633 and U.S. Patent No. 4,891,211, incorporated
30 herein by reference.

Several patents and patent applications disclose the use of abrasive materials in solid, oral compositions, see for example U.S. Patent Nos. 5,147,632 and 5,496,541, EP Patent No. 372,603, International Publication Nos. WO 02/19834 and WO 01/56399.

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US Patent Application US 2002/0142068 discloses chewing gum formulations including sodium pyrophosphate and encapsulated aspartame.

US Patent No. 4,233,288 discloses a gum emulsified liquid composition for delivering and preserving the liquid content in the mouth. The herein disclosed examples describe a gum emulsified liquid composition comprising 5% of calcium pyrophosphate and more than 50% of liquid components.

U.S. Patent No. 3,590,120 discloses a chewing gum comprising a polishing agent comprising a mixture of fine and coarse zirconium silicate particles. Disclosed herein are reference chewing gum compositions containing 10% of calcium pyrophosphate or 10% of calcium carbonate, respectively. The cleaning/polishing effects of said compositions are shown to decrease in the order ZrSiO_4 , CaCO_3 and CaP_2O_7 . Indicative studies have however shown some problematic toxicological properties of zirconium silicate (Elmore AR, Cosmetic Ingredient Review Expert Panel, Final report on the safety assessment of aluminum silicate, calcium silicate, magnesium aluminium silicate, magnesium silicate, magnesium trisilicate, sodium magnesium silicate, zirconium silicate, attapulgite, bentonite, Fuller's earth, hectorite, kaolin, lithium magnesium silicate, lithium magnesium sodium silicate, montmorillonite, pyrophyllite, and zeolite, Int. J Toxicol. 2003; 22 Suppl. 1:37-102) and the use of zirconium compounds in solid oral compositions is prohibited in a number of countries.

Thus, a need exists in the art to identify and use abrasives in solid, oral tooth whitening compositions, to obtain an increased whitening effect, which compositions are safe and convenient to administer and apply.

Disclosure of Invention

The present invention relates to a solid oral tooth whitening confectionary composition comprising more than 75% by weight of solid materials, said composition comprising:

- a) a confectionary base,
- b) conventional confectionary additives,
- c) a tooth whitening agent comprising an alkaline or alkaline earth metal pyrophosphate.

Furthermore, the present invention relates to the use of a composition according to the present invention to whiten tooth surfaces.

Furthermore, the present invention relates to a method of whitening tooth surfaces.

Brief description of drawing

The invention is explained in greater detail below with reference to the accompanying drawing, in which

Fig. 1 is a diagrammatic view of the process for preparing hard-boiled lozenges as disclosed in example 1.

Best Modes for Carrying out the Invention

In a preferred embodiment according to the present invention the solid oral tooth whitening confectionary composition comprises confectionary additives and a tooth whitening agent comprising calcium pyrophosphate.

The use of abrasives, among which calcium pyrophosphate is normally classified, in

confectionary, would not be expected to cause a tooth whitening effect comparable to the effect when used in compositions intended for continuous chewing, i.e. chewing gum compositions, since confectionary compositions are not chewed to the same extent. As a consequence the stain removal effect of the abrasive material would be expected to be insignificant due to the minimal mechanical rubbing on the tooth surface.

It has now surprisingly been demonstrated, using *in vitro* tests, that improved stain removal effects of solid, oral confectionary compositions can be achieved using from 0.1 to 10 % of calcium pyrophosphate as an abrasive agent, compared to the use of the often used abrasive, calcium carbonate.

These effects are further unexpected, seen in the light of the results presented in US 3,590,120, in which the effect of CaP_2O_7 was significantly poorer than that of CaCO_3 .

Typically, confectionary tooth whitening compositions are intended to comprise a recommended daily dose of tooth whitening agent of about 4 to 700 mg. Conveniently, this dose may be divided into multiple sub-doses, which can be ingested as needed such as at suitable intervals or in suitable situations, i.e. after meals, after use of substances known to induce stain formation, i.e. coffee, tea, red wine, tobacco and the like. Assuming a confectionary unit weight of 1300 mg, a content of 0.1% to 10% of tooth whitening agent corresponds to a composition unit content of approximately 1.3 to 130 mg of tooth whitening agent.

The compositions of the invention are essentially solid and comprise more than 75%, preferably more than 85%, even more preferably more than 95%, by weight of the composition of solid materials.

The solid material comprises a confectionary base, conventional confectionary additives and a tooth whitening agent.

In a preferred embodiment of the invention, calcium pyrophosphate is present in an amount of between 0.5% and 9%, preferably between 1.0% and 6.5 %, even more preferably between 1.5% and 4.0 %, by weight of the composition.

5

In a preferred embodiment, the compositions of the invention are formulated as confectionary compositions comprising a confectionary base, said confectionary base preferably comprising from 5% to 99%, particularly from 15% to 98%, preferably 30% to 97% by weight of the composition. Non-limiting examples of confectionary
10 compositions according to the invention include hard-boiled, grained sugar confectionary, lozenges, chocolate, compressed tablets, gummy confectionary and jellies. Confectionary base materials are well known to the person skilled in the art and vary according to the type of confectionary composition, e.g. the compositions mentioned above.

15

The compositions according to the present invention may contain one or more conventional ingredients such as sweeteners, high intensity sweeteners, taste enhancers, flavouring agents and the like. Sweeteners, high intensity sweeteners and taste enhancers are well known to the skilled person. Non-limiting examples of sweeteners
20 comprise sugar sweeteners including saccharides such as sucrose, dextrose, glucose, maltose, dextrins, D-tagatose, trehalose, dried invert sugar, fructose, levulose, galactose, corn syrup solids, and the like, alone or in combination. Other examples of sweeteners comprise sugarless sweeteners including polyhydric alcohols such as sorbitol, mannitol, xylitol, glycerol, hydrogenated starch hydrolysates, maltitol,
25 isomaltitol, erythritol, lactitol and the like, alone or in combination. Sugarless sweeteners are preferred.

Preferred high intensity sweeteners include but are not limited to sucralose, aspartame, superaspartame, sucronic acid, twinsweet™, neohesperidin dihydrochalcone,
30 stevia, brazzein, mogroside, monatin™, ajinomoto™ sweetener, alapyridaine, tagatose, rebaudioside A, salts of acesulfame, alitame, saccharin or salts thereof, neotame,

cyclamic acid and salts thereof, glycyrrhizin, dihydrochalcones, thaumatin, mon-nelin, sterioside and the like, alone or in combination.

A variety of flavours known in the art may be used, such as cinnamon, wintergreen,
5 eucalyptus, spearmint, peppermint, menthol, anise as well as fruit flavours such as apple, pear, peach, strawberry, cherry, apricot, orange, watermelon, banana and the like; bean-derived flavours, such as coffee, cocoa and the like. Flavouring agents are incorporated in the confectionary formulation at a concentration of about 0.1 to about 5 % by weight and preferably 1 to 3 % by weight.

10

The compositions of the invention may or may not contain sugar. Sugar-free compositions, however, are preferred.

It may be advantageous to include one or more additional tooth whitening agents.

15

Examples of such additional tooth whitening agents are well known in the art and include abrasives as well as bleaching agents. Abrasive materials comprise as non-limiting examples silica, alumina, calcium carbonate, dicalcium phosphate, hydroxyapatite, trimetaphosphates and insoluble hexametaphosphates. Bleaching
20 agents comprise agents such as peroxy compounds, e.g. potassium peroxydiphosphate and urea-peroxid. Effervescing systems such as sodium bicarbonate, alone or in combination with citric acid as well as colour change systems may also be incorporated into compositions according to the present invention.

25 In the compositions according to the invention, said additional whitening agents are usually present in between 0.01% and 5.0%, preferably between 0.05 and 1.0%, more preferably between 0.1% and 0.5% by weight of the composition.

A preferred additional tooth whitening agent comprises a bicarbonate salt. In one
30 embodiment, said bicarbonate salt comprises sodium bicarbonate in an amount of between 0.1% and 0.5% by weight of the compositions.

A range of active agents may be added to the compositions of the invention. Such agents may comprise one or more of the following; oral hygiene promoting agents, anti-calculus agents, anti-microbial agents, anti-inflammatory agents, desensitising agents, therapeutically active agents, remineralising agents. Non-limiting examples
5 comprise anti-carries agents such as sodium, calcium, magnesium and stannous fluoride, amine fluorides, disodium monofluorophosphate, sodium trimetaphosphate and casein; antimicrobial agents, e.g. Triclosan, chlorhexidine, copper, zinc and stannous salts such as zinc citrate, zinc sulphate, zinc glycinate, sodium zinc citrate and stan-
10 nous pyrophosphate, sanguinarine extract, metronidazole, quaternary ammonium compounds, such as cetylpyridinium chloride; bis-guanides, such as chlorhexidine digluconate, hexetidine, octenidine, alexidine; and halogenated bisphenolic compounds, such as 2,2' methylenebis-(4-chloro-6-bromophenol); anti-inflammatory agents such as ibuprofen, flurbiprofen, aspirin, indomethacin etc.; plaque acid buff-
15 ers such as urea, calcium lactate, calcium glycerophosphate and strontium polyacrylates; desensitising agents, e.g. potassium citrate, potassium chloride, potassium tartrate, potassium bicarbonate, potassium oxalate, potassium nitrate and strontium salts; anti-calculus agents, e.g. hypophosphite-containing polymers, organic phosphonates and phosphocitrates etc.; gum protection agents, e.g. vegetable oils such as
20 sunflower oil, rape seed oil, soybean oil, safflower oil; silicone oil; and hydrocarbon oil; pharmaceutically acceptable carriers, e.g. starch, sucrose, water or water/alcohol systems etc.; surfactants, such as anionic, nonionic, cationic and zwitterionic or amphoteric surfactants. Other agents which may be incorporated in the compositions of the present invention are agents to counter breath malodour and include water solu-
25 ble zinc salts (at least 1% soluble) particularly zinc chloride, zinc acetate, zinc citrate and zinc gluconate.

The additives, the whitening agents and the optional active agents comprised by the present invention may be encapsulated. This may be done in order to achieve a slow
30 release of the encapsulated agents upon entering the oral environment. For example a longer lasting sweetening of the compounds comprised by the present invention

may be achieved by encapsulating the sweetening agents. A longer release time of the whitening agents as well as any therapeutic compound may likewise be achieved.

- 5 Another advantage of encapsulating the agents comprised by the invention may be to obtain an increased stability of the agents, thus lending a longer storage life at a greater range of storage conditions to the compositions of the invention.

- 10 Any standard method giving partial or full encapsulation can be used for encapsulation. Suitable methods include, but are not limited to, spray drying, spray chilling, fluid-bed coating, and coacervation. These methods can be used individually or in any combination in a single step process or multiple step process.

- 15 Generally, compositions of high organic solubility, good film forming properties, and low water solubility, provide a suitable encapsulation. These compositions include acrylic polymers and copolymers, carboxyvinyl polymers, polyamides, polystyrene, polyvinyl acetate, polyvinyl acetate phthalate, polyvinyl pyrrolidone, and waxes.

- 20 However, only food grade materials should be used for the encapsulation. Two standard food grade coating materials, which are good formers, but not water soluble, are shellac and Zein. Others which are more water soluble, but also good film formers, are materials such as agar, alginates, a wide range of cellulose derivatives like ethyl cellulose and hydroxypropylmethyl cellulose, dextrin, gelatin and modified
25 starches. It is also possible to use other encapsulants like acacia or maltodextrin for encapsulation.

- In yet another embodiment of the invention, it may be desirable to include a supplement, such as vitamins and/or minerals in the composition according to the
30 invention. Vitamins are preferably added in concentrations of between 10 % - 100% of the recommended daily allowance (RDA).

Especially vitamin C may be added to the compositions of the invention.

It may be desirable to include urea in the compositions of the invention. Urea may
5 be added as a plaque acid neutralising agent. Usually urea is added to the compositions in between 0.1% and 25%, particularly between 0.4% and 10%, preferably between 0.6% and 5.0%, more preferably between 0.7 % and 3.5%, even more preferably between 0.8 % and 2.5% by weight.

10 The compositions according to the invention may be in the form of lozenges. In a particularly embodiment of the invention said lozenges are hard-boiled lozenges. Lozenges may be prepared according to conventional procedures as disclosed in more detail in the examples below. Thus compositions according to the invention in the form of lozenges may be in the form of plain lozenges, wherein all ingredients
15 are mixed more or less homogenously. Furthermore the compositions according to the invention may be in the form of lozenges having either a powder-filled or a liquid-filled centre.

In another embodiment, the present invention relates to the use of the compositions
20 of the invention to whiten tooth surfaces and/or prevent discolouration of tooth surfaces. Especially, the compositions of the invention may be used to remove or prevent discolouration of teeth due to the use of tobacco-related products and/or consumption of red wine or related products as well as coffee, tea or related products.

25 Examples

Example 1:

Manufacture of sugar-free hard-boiled compositions containing calcium pyrophos-
30 phate

The manufacturing process is diagrammatically shown in Fig. 1.

Base composition:

	Water content	2 %
5	Isomalt	98 %
	Acesulfame K	0.05 %
	Flavours / Colours	QS

A confectionary base containing 98% isomalt, 0.05 % acesulfame K and approx. 2
10 % water was prepared by dissolving isomalt and acesulfame K in water followed by
cooking. The mixture was thereafter placed under vacuum. Colouring agents and
flavouring agents are added after vacuum treatment. Optionally several vacuum
chambers can be used. The composition was subsequently cooled until the composi-
15 tion reached a temperature suitable to form the desired confectionary pieces. There-
after it was formed and subjected to a final cooling step.

Composition 1 – plain sweet:

	Water content	2 %
20	Isomalt	92 %
	Polyol syrup	4%
	Calcium pyrophosphate	2%
	Acesulfame K	0.05 %

25 Calcium pyrophosphate in an amount corresponding to 2% of the total composition
was added to the above confectionary base together with a polyol syrup (4% of to-
tal). Addition was done after the cooking of the confectionary base during the vac-
uum process.

30 Composition 2 – powder-filled:

Powder Filling recipe

Polyols spray	70 %
Calcium Pyrophosphate	30 %
Flavours Colours	QS

5

Total recipe

Base SF Hard Boiled	93.3 %
Powder filling	6.7 %

- 10 A powder-filled confectionary composition was prepared by adding a total of 6.7% of a powder comprising 30% calcium pyrophosphate and 70% of a polyol spray to 93.3% of the base confectionary composition during the forming step in the above procedure. The final composition contained approx. 2% calcium pyrophosphate.

- 15 Composition 3 – liquid-filled:

Liquid Filling

Water Content	10 %
Polyols spray	50%
20 Polyols syrup	26 %
Calcium Pyrophosphate	14 %
Flavours Colours	QS

Total recipe

25 Base SF Hard Boiled	85 %
Powder filling	15 %

- 30 A liquid-filled confectionary composition was prepared by adding a total of 15% of a liquid comprising 14% calcium pyrophosphate, 50% of a polyol spray, 26% of a polyol syrup and 10 % water to 85% of the base confectionary composition during the forming step in the above procedure. The final composition contained approx.

2% calcium pyrophosphate

Example 2:

- 5 The effect of solid oral tooth whitening confectionary compositions on the removal of extrinsic stains from teeth after 60 minutes treatment.

In the following non-limiting example, the inventive confectionary compositions are formulated as standard lozenge compositions, wherein the sweeteners constitute the
10 confectionary base. In addition to the specified tooth whitening ingredients the standard lozenge compositions used herein consists essentially of the following ingredients:

	Isomalt	96.22 %-99.67%
15	Aspartame	0.09%
	Acesulfame K	0.04%-0.05%
	Xylitol	0.00 - 1.29%
	Menthol Flavour	0.03%-0.04%
	Lemon Flavour	0.15 %

20

The above lozenge composition was prepared similar to the procedure of example 1, composition 1.

- 25 An assay for the effect of the tooth whitening compositions on the removal of extrinsic stains on tooth surfaces was set up. The assay consisted of treatments with five different lozenge compositions and a treatment with distilled water.

The lozenge compositions contained by weight of the composition the confectionary additives stated above, in addition to one of the following combinations;

30

- 1) 1.44 % of calcium carbonate and 0.12 % sodium bicarbonate (CaCO_3 + Na-

- HCO₃),
- 2) 2.08 % of calcium pyrophosphate and 0.12 % sodium bicarbonate (CaP₂O₇ + NaHCO₃),
- 3) 0.12 % sodium bicarbonate (NaHCO₃),
- 5 4) 2.08 % of calcium pyrophosphate (CaP₂O₇),
- 5) No abrasive (negative control),
- 6) No lozenge (negative control, water).

The experiments were conducted using a modification of the laboratory method described by Stookey, GK: Burkhart, T.A: and Schemehorn, B.R; In vitro removal of stain with dentifrices, J Dent Res 61(11):1236-1239, Nov 1982, which has been shown to correlate with the cleaning/whitening properties of dentifrices in clinical trials. The amount of stain on the teeth before and after treatment is measured quantitatively using a colorimeter.

15

Each composition was tested on four enamel pieces containing a stained pellicle layer. The enamel pieces were stained with a broth comprising coffee, tea, red wine, gastric mucin as well as a culture of the micro-organism *Micrococcus luteus*. Prior to treatment the colour of the stained teeth was measured using a calorimeter, and 6 groups containing 4 teeth each balanced for baseline stains were formed.

The lozenge compositions were diluted 1 part by weight with 3 parts distilled water to simulate salivary dilution. 10 ml of lozenge test solution was placed in a beaker along with one of the stained teeth and mechanically stirred for 20 minutes. Each tooth received a total of 60 minutes treatment divided into three 20-minute treatments. Each new 20 minute treatment was carried out using fresh lozenge test solution.

After completion of treatments the stains remaining on the teeth were measured using a colorimeter. The before and after measurements were used to calculate the overall change in colour (ΔE), based on the standard CIELAB colour procedure.

Hereafter teeth were cleaned completely using professional dental cleaning equipment, and another measurement using the colorimeter was done in order to determine the total amount of removable stain initially present on the teeth. A percent stain reduction was then calculated by dividing the amount of stain removed by the
 5 lozenge treatments by the total amount of removable stain.

The mean values and the standard deviations for each treatment group are shown in table 1, column 2 (ΔE). The maximum removal and the concomitant standard deviations are shown in column 3 (Maximum ΔE). The % of reduction of stains is shown
 10 in column 4 (Reduction).

The values in each column with the same superscript are not statistically different, while those with different superscript are different at $p < 0.05$ based on ANOVA and SNK testing.

15

Table 1

Composition	ΔE	Maximum ΔE	Reduction
1 ($\text{CaCO}_3 + \text{NaHCO}_3$)	0.69 ± 0.37^a	27.58 ± 2.12^a	2.5% ^a
2 ($\text{CaP}_2\text{O}_7 + \text{NaHCO}_3$)	2.83 ± 0.67^b	30.72 ± 1.04^a	9.2% ^b
3 (NaHCO_3)	0.70 ± 0.12^a	29.88 ± 2.32^a	2.3% ^a
4 (CaP_2O_7)	5.62 ± 1.66^c	31.56 ± 3.57^a	17.6% ^c
5 (no abrasive)	0.59 ± 0.35^a	28.59 ± 0.66^a	2.1% ^a
6 (water)	0.51 ± 0.24^a	30.63 ± 3.40^a	1.7% ^a

Claims

1. A solid oral tooth whitening confectionary composition comprising more than 75% by weight of solid materials, said composition comprising:
 - 5 a) a confectionary base,
 - b) conventional confectionary additives,
 - c) a tooth whitening agent comprising an alkaline or alkaline earth metal pyrophosphate.
- 10 2. A composition according to claim 1 in which said a tooth whitening agent comprises calcium pyrophosphate.
3. A composition according to claim 2 in which said calcium pyrophosphate is present in an amount of between 0.1 and 10% by weight of the composition.
- 15 4. The composition according to claim 3 in which said calcium pyrophosphate is present in an amount of between 0.5% and 9%, preferably between 1.0 % and 6.5 %, even more preferably between 1.5 % and 4.0 %, by weight of the composition.
- 20 5. The composition according to any of the preceding claims in which said conventional confectionary ingredients comprise one or more of the following: sweeteners, high intensity sweeteners, taste enhancers, flavouring agents, colouring agents.
- 25 6. The composition according to any of the preceding claims in which said composition is essentially sugar-free.
7. The composition according to any of the preceding claims comprising one or more additional tooth whitening agents.
- 30 8. The composition according to claim 7 in which said additional tooth whitening agent(s) is/are present in between 0.01% and 5.0%, more particularly between 0.05

and 1.0%, most preferably between 0.1% and 0.5% by weight of the composition.

9. The composition according to claim 7 or 8 in which said additional tooth whitening agent comprises a bicarbonate salt.

10. The composition according to claim 9 in which said additional tooth whitening agent comprises sodium bicarbonate, said agent being present in between 0.1% and 0.5% by weight of the composition.

11. The composition according to any of the preceding claims in which said additives and/or tooth whitening agents are encapsulated.

12. The composition according to any of the preceding claims further comprising one or more of the following: oral hygiene promoting agents, anti-calculus agents, anti-microbial agents, anti-inflammatory agents, desensitising agents, therapeutically active agents, remineralising agents.

13. The composition according to any of the preceding claims further comprising a supplement.

14. The composition according to claim 13 in which said supplement comprises vitamin C.

15. The composition according to claim 12 in which the oral hygiene promoting agent comprises urea, said urea being present in between 0.1% and 25%, particularly between 0.4% and 10%, preferably between 0.6% and 5.0%, more preferably between 0.7 % and 3.5%, even more preferably between 0.8 % and 2.5% by weight.

16. The composition according to any of the preceding claims in the form of lozenges.

17. The composition according to claim 16 in the form of hard-boiled lozenges.

18. A use of a composition according to any of the preceding claims to whiten tooth surfaces.

5

19. A use of a composition according to claims 1-17 to whiten tooth surfaces, said tooth surfaces being discoloured after use of red wine or related products.

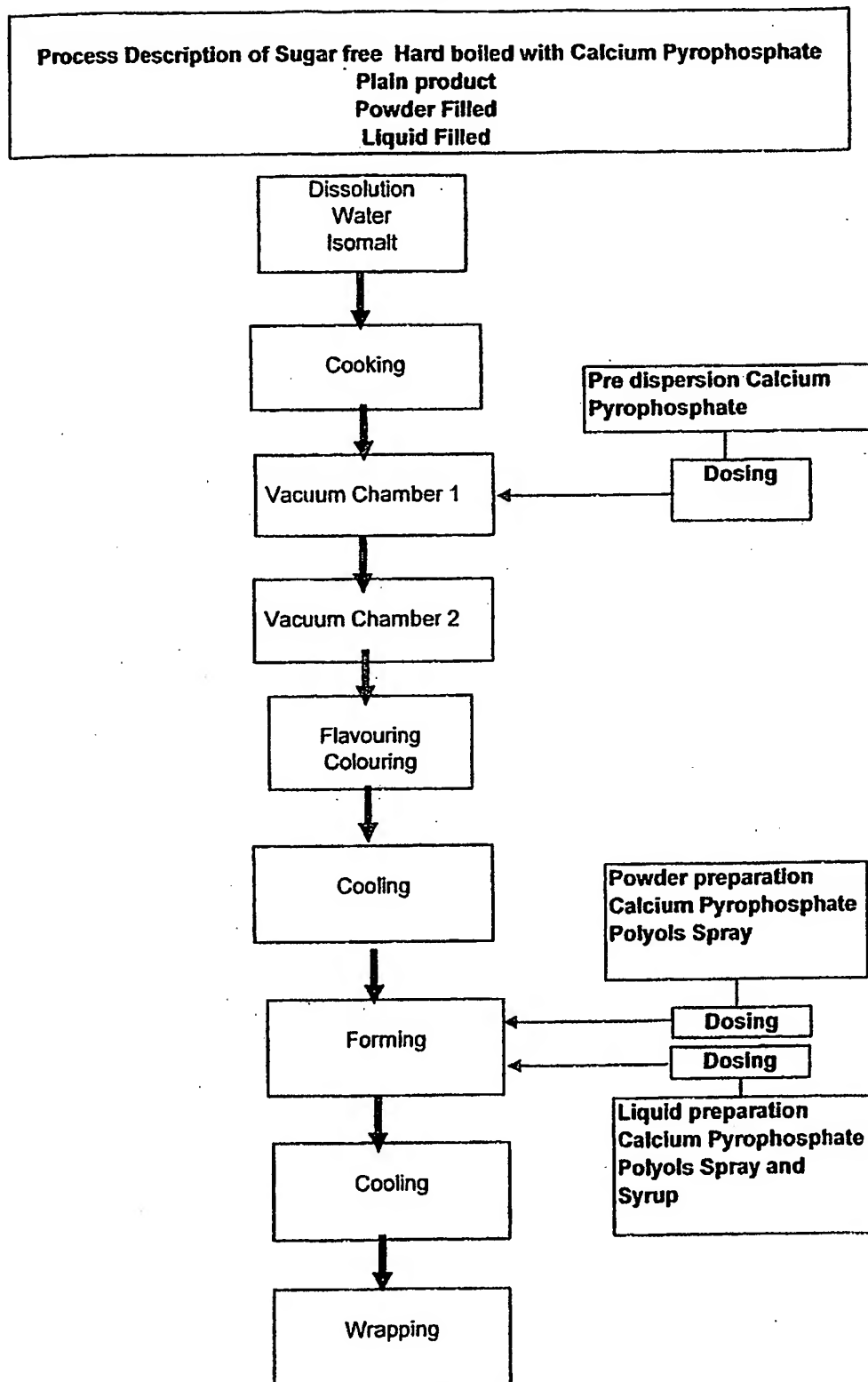
20. A use of a composition according to any of the claims 1-17 to whiten tooth surfaces, said tooth surfaces being discoloured after use of coffee-related products.

21. A method of whitening tooth surfaces by consuming a solid, oral tooth whitening confectionary composition according to any of claims 1-17.

15 22. A method of whitening tooth surfaces by consuming a solid oral tooth whitening confectionary composition according to claims 1-17, said tooth surfaces being discoloured after use of red wine or related products.

20 23. A method of whitening tooth surfaces by consuming a solid oral tooth whitening confectionary composition according to any of the claims 1-17, said tooth surfaces being discoloured after use of coffee-related products.

Fig 1.



INTERNATIONAL SEARCH REPORT

International Application No.

PCT/EP2004/013963

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 A61K7/16 A23G3/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 A61K A23G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
X	WO 03/002056 A (CHURCH & DWIGHT CO., INC) 9 January 2003 (2003-01-09) page 1, line 5 - line 6 page 5, line 15 - line 16 page 13, line 1 - line 5 page 14, line 5 - line 9 claims	1-13, 16-23
X	US 2003/072841 A1 (RAJAJIAH JAYANTH ET AL) 17 April 2003 (2003-04-17) paragraphs '0032!', '0038!', '0059!', '0071! claims	1-23



Further documents are listed in the continuation of box C



Patent family members are listed in annex

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INTERNATIONAL SEARCH REPORT

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PCT/EP2004/013963

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
X	WO 03/017964 A (UNILEVER N.V; UNILEVER PLC; HINDUSTAN LEVER LTD) 6 March 2003 (2003-03-06) page 5, line 6 page 6, line 11 - line 22 page 7, line 13 - line 15 page 8, line 14 - line 15 claim 14	1-17
Y	-----	18-23
X	WO 99/12517 A (SMITHKLINE BEECHAM CORPORATION; SMITHKLINE BEECHAM PLC; CASH, MICHAEL;) 18 March 1999 (1999-03-18) page 1, line 1 - line 33 page 9, lines 3,25 - line 26 page 10, line 2 - line 3 claims	1-13, 16-23
Y	-----	18-23
X	US 2003/072722 A1 (NATHOO SALIM A) 17 April 2003 (2003-04-17) paragraphs '0002!', '0005!', '0043! table 1 claims 1,10	1,18-23
X	DE 36 45 147 C2 (COLGATE-PALMOLIVE CO., NEW YORK) 9 November 2000 (2000-11-09) the whole document	1,5,6,9, 12,13, 16-23
X	GB 1 018 665 A (UNILEVER LIMITED) 26 January 1966 (1966-01-26) the whole document	1-8,12

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No.

PCT/EP2004/013963

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 03002056	A	09-01-2003	US 6355229 B1	12-03-2002
			WO 03002056 A2	09-01-2003
US 2003072841	A1	17-04-2003	CA 2441536 A1	26-09-2002
			CN 1520261 A	11-08-2004
			EP 1370152 A2	17-12-2003
			HU 0303471 A2	01-03-2004
			JP 2004520062 T	08-07-2004
			MX PA03008495 A	08-12-2003
			PL 364282 A1	13-12-2004
			WO 02074099 A2	26-09-2002
			CA 2441865 A1	26-09-2002
			EP 1370227 A2	17-12-2003
			HU 0303498 A2	01-03-2004
			JP 2004524334 T	12-08-2004
			MX PA03008497 A	08-12-2003
			WO 02074274 A2	26-09-2002
			WO 02074276 A2	26-09-2002
			US 2002187108 A1	12-12-2002
			US 6500406 B1	31-12-2002
			US 2003086878 A1	08-05-2003
			US 2003082113 A1	01-05-2003
WO 03017964	A	06-03-2003	DE 10238534 A1	19-02-2004
			DE 10238535 A1	15-05-2003
			DE 10238537 A1	26-06-2003
			DE 10238538 A1	22-05-2003
			WO 03017962 A1	06-03-2003
			WO 03017963 A1	06-03-2003
			WO 03017964 A1	06-03-2003
			WO 03017965 A1	06-03-2003
			EP 1418882 A1	19-05-2004
			EP 1418883 A1	19-05-2004
			FR 2828805 A1	28-02-2003
			FR 2828806 A1	28-02-2003
			FR 2828807 A1	28-02-2003
			FR 2828808 A1	28-02-2003
			GB 2380405 A	09-04-2003
			GB 2380406 A	09-04-2003
			GB 2380407 A	09-04-2003
			GB 2380408 A	09-04-2003
			US 2003068282 A1	10-04-2003
			US 2003068283 A1	10-04-2003
			US 2003077232 A1	24-04-2003
			US 2003082112 A1	01-05-2003
WO 9912517	A	18-03-1999	AU 9300998 A	29-03-1999
			WO 9912517 A1	18-03-1999
			ZA 9808191 A	09-03-1999
US 2003072722	A1	17-04-2003	US 2005019276 A1	27-01-2005
DE 3645147	C2	09-11-2000	US 4627977 A	09-12-1986
			US 4806340 A	21-02-1989
			AR 243372 A1	31-08-1993
			AT 406015 B	25-01-2000
			AT 237586 A	15-05-1989
			AT 406016 B	25-01-2000

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP2004/013963

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
DE 3645147	C2	AT 237686 A	15-05-1989
		AU 594703 B2	15-03-1990
		AU 6204986 A	19-03-1987
		BE 905428 A1	12-03-1987
		BE 905429 A1	12-03-1987
		BR 8604377 A	12-05-1987
		CA 1332359 C	11-10-1994
		CA 1275937 C	06-11-1990
		CA 1339410 C	02-09-1997
		CH 668907 A5	15-02-1989
		CH 668908 A5	15-02-1989
		DE 3629503 A1	26-03-1987
		DE 3629504 A1	26-03-1987
		DK 433386 A	14-03-1987
		DK 433486 A	14-03-1987
		EG 18045 A	28-02-1993
		ES 2003096 A6	16-10-1988
		ES 2013787 A6	01-06-1990
		FI 863708 A ,B,	14-03-1987
		FI 863709 A ,B,	14-03-1987
		FR 2587211 A1	20-03-1987
		GB 2180157 A ,B	25-03-1987
		GB 2182244 A ,B	13-05-1987
		GB 2211738 A	12-07-1989
		GR 862312 A1	19-01-1987
		GR 862314 A1	29-01-1987
		HK 26793 A	02-04-1993
		HK 53293 A	11-06-1993
		IE 59557 B1	09-03-1994
		IE 59532 B1	09-03-1994
		IL 79892 A	21-06-1992
		IL 79893 A	25-05-1992
		IL 94197 A	21-06-1992
		IL 96686 A	25-05-1992
		IN 167015 A1	18-08-1990
		IT 1196621 B	16-11-1988
		IT 1196622 B	16-11-1988
		JP 8018961 B	28-02-1996
		JP 62096409 A	02-05-1987
		JP 1826372 C	28-02-1994
		JP 5030803 B	11-05-1993
		JP 62111911 A	22-05-1987
		KR 9306345 B1	14-07-1993
		KR 9311550 B1	11-12-1993
GB 1018665	A	26-01-1966	BE 625382 A
			CA 974453 A1
			CH 446617 A
			DE 1250970 B
			FR 1347921 A
			IT 943006 B
			LU 42774 A1
			NL 149370 B
			NL 286101 A

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